

Introduction to Chemistry

Module Information

2022.02, Approved

Summary Information

| Module Code | 3418FNDSCI |
|---------------------|----------------------------------|
| Formal Module Title | Introduction to Chemistry |
| Owning School | Pharmacy & Biomolecular Sciences |
| Career | Undergraduate |
| Credits | 20 |
| Academic level | FHEQ Level 3 |
| Grading Schema | 40 |

Teaching Responsibility

| LJMU Schools involved in Delivery | |
|-----------------------------------|--|
| Pharmacy & Biomolecular Sciences | |

Learning Methods

| Learning Method Type | Hours |
|----------------------|-------|
| Lecture | 33 |
| Online | 10 |
| Practical | 8 |
| Workshop | 6 |

Module Offering(s)

| Display Name | Location | Start Month | Duration Number Duration Unit |
|--------------|----------|-------------|-------------------------------|
| SEP-CTY | СТҮ | September | 12 Weeks |

Aims and Outcomes

After completing the module the student should be able to:

Learning Outcomes

| Code | Number | Description |
|------|--------|--|
| MLO1 | 1 | Use laboratory data to draw appropriate conclusions. |
| MLO2 | 2 | Perform calculations using parameters that measure properties of matter. |
| MLO3 | 3 | Describe sub-atomic structure and apply atomic theory to interpret properties of matter. |
| MLO4 | 4 | Relate reaction rates and physicochemical changes to fundamental scientific laws. |

Module Content

| Outline Syllabus | Accuracy, precision and equipment errors in scientific measurement. Atomic theory, including electron sublevels and radioactivity. Molecular bonding and intermolecular bonding. The periodic table and an overview of trends. Moles, mass, molarity and the gas laws. Relative atomic mass and its determination using a simple mass spectrometer. Structure and properties of gases, liquids and solids. Origins of acid and base behaviour and chemical equilibria. Spontaneous reactions, free energy, enthalpy and entropy. Rate of reactions, rate laws, reaction order and half-life. Collision theory and factors affecting rate. Nomenclature and structure of common organic functional groups. |
|------------------------|---|
| Module Overview | |
| Additional Information | Covers introductory laboratory skills and fundamental physical and inorganic chemistry concepts alongside a brief introduction to organic chemistry. |

Assessments

| Assignment Category | Assessment Name | Weight | Exam/Test Length (hours) | Module Learning Outcome Mapping |
|---------------------|-----------------|--------|--------------------------|------------------------------------|
| Report | Practical | 50 | 0 | MLO1, MLO2 |
| Centralised Exam | MCQ Exam | 50 | 1.5 | MLO4, MLO2, MLO3 |

Module Contacts

Module Leader

| Contact Name | Applies to all offerings | Offerings |
|---------------|--------------------------|-----------|
| Philip Denton | Yes | N/A |

Partner Module Team

| Contact Name | Applies to all offerings | Offerings |
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