

Module Proforma

Approved, 2022.03

Summary Information

| Module Code | 3458FNDSCI |
|---------------------|----------------------------------|
| Formal Module Title | Introduction to Chemistry |
| Owning School | Pharmacy & Biomolecular Sciences |
| Career | Undergraduate |
| Credits | 20 |
| Academic level | FHEQ Level 3 |
| Grading Schema | Pass/Not Pass |

Module Contacts

Module Leader

| Contact Name | Applies to all offerings | Offerings |
|---------------|--------------------------|-----------|
| Philip Denton | Yes | N/A |

Module Team Member

| Contact Name | Applies to all offerings | Offerings | |
|----------------|--------------------------|-----------|--|
| Barry Nicholls | Yes | N/A | |

Partner Module Team

Teaching Responsibility

| LJMU Schools involved in Delivery | |
|-----------------------------------|--|
| Pharmacy & Biomolecular Sciences | |

Learning Methods

| Learning Method Type | Hours |
|----------------------|-------|
| Lecture | 33 |
| Online | 10 |
| Practical | 8 |
| Workshop | 6 |

Module Offering(s)

| Offering Code | Location | Start Month | Duration |
|---------------|----------|-------------|----------|
| SEP-CTY | CTY | September | 12 Weeks |

Aims and Outcomes

| Aims | To provide student with a grounding in laboratory skills, fundamental physical and inorganic chemistry concepts, and an introduction to organic chemistry. |
|------|--|
|------|--|

Learning Outcomes

After completing the module the student should be able to:

| Code | Description |
|------|--|
| MLO1 | Use laboratory data to draw appropriate conclusions. |
| MLO2 | Perform calculations using parameters that measure properties of matter. |
| MLO3 | Describe sub-atomic structure and apply atomic theory to interpret properties of matter. |
| MLO4 | Relate reaction rates and physicochemical changes to fundamental scientific laws. |

Module Content

Outline Syllabus

Accuracy, precision and equipment errors in scientific measurement. Atomic theory, including electron sublevels and radioactivity. Molecular bonding and intermolecular bonding. The periodic table and an overview of trends. Moles, mass, molarity and the gas laws. Relative atomic mass and its determination using a simple mass spectrometer. Structure and properties of gases, liquids and solids. Origins of acid and base behaviour and chemical equilibria. Spontaneous reactions, free energy, enthalpy and entropy. Rate of reactions, rate laws, reaction order and half-life. Collision theory and factors affecting rate. Nomenclature and structure of common organic functional groups.

Module Overview

Additional Information

Module covers introductory laboratory skills and fundamental physical and inorganic chemistry concepts alongside a brief introduction to organic chemistry. This is a core module for all Level 3 students on Pharmacy with a foundation year. This module is Pass/Fail with a minimum 55% requirement in each assessed item.

Assessments

| Assignment Category | Assessment Name | Weight | Exam/Test Length (hours) | Learning Outcome Mapping |
|---------------------|-----------------|--------|--------------------------|--------------------------------|
| Centralised Exam | MCQ Exam | 50 | 1.5 | MLO2, MLO1 |
| Report | Practical | 50 | 0 | MLO4, MLO3, MLO2 |